

# Formal Total Synthesis of Fostriecin via 1,4-Asymmetric Induction Using Cobalt-Alkyne Complex

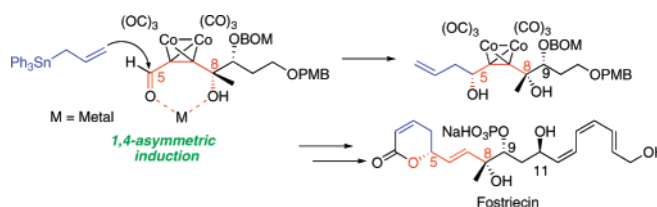
Yujiro Hayashi,\* Hirofumi Yamaguchi, Maya Toyoshima, Kotaro Okado, Takumi Toyo, and Mitsuru Shoji

Department of Industrial Chemistry, Faculty of Engineering, Tokyo University of Science, Kagurazaka, Shinjuku-ku, Tokyo 162-8601, Japan

hayashi@ci.kagu.tus.ac.jp

Received January 28, 2008

## ABSTRACT



The synthesis of a protected dephosphofostriecin, and thereby a formal synthesis of fostriecin, has been accomplished. Two of the four chiral centers are controlled by an external chiral auxiliary and the other two are synthesized stereoselectively, one by a novel 1,4-asymmetric induction using cobalt-alkyne complex, and the other by 1,3-asymmetric induction.

Fostriecin (**1**, CI-920), a novel secondary metabolite of *Streptomyces pulveraceus*, is a selective inhibitor of protein phosphatase 2A (PP2A) that displays antitumor activity against a diverse panel of tumor cell lines and in vivo activity

toward lymphoid leukemias.<sup>1</sup> This important biological activity has attracted the attention of many synthetic chemists. Both the absolute and the relative stereochemistry were determined by Boger,<sup>2</sup> who also accomplished the first total synthesis in 2001.<sup>3</sup> Since then, several excellent asymmetric syntheses have appeared,<sup>4</sup> as well as a number of synthetic studies.<sup>5</sup>

Structurally fostriecin has four chiral centers and contains a conformationally labile *cis-cis-trans* triene moiety. Shibasaki<sup>4g,i,k</sup> and Falck<sup>4b</sup> constructed all the stereocenters using four different catalytic asymmetric reactions, while three of the four chiral centers were constructed using asymmetric catalytic reactions by Jacobsen,<sup>4a</sup> Imanishi,<sup>4c,d</sup> Kobayashi,<sup>4e</sup> and Trost.<sup>4h</sup> In these syntheses, asymmetric catalytic reactions have been utilized at least twice. Yadav employed D-glucose as a chiral starting material,

(1) (a) Lewy, D. S.; Gauss, C. M.; Soenen, D. R.; Boger, D. L. *Curr. Med. Chem.* **2002**, *9*, 2005. (b) McCluskey, A.; Sim, A. T. R.; Sakoff, J. A. *J. Med. Chem.* **2002**, *45*, 1151. (c) DeJong, R. S.; Mulder, N. H. *Anti-Cancer Drugs* **1997**, *8*, 413.

(2) Boger, D. L.; Hirota, M.; Lewis, B. M. *J. Org. Chem.* **1997**, *62*, 1748.

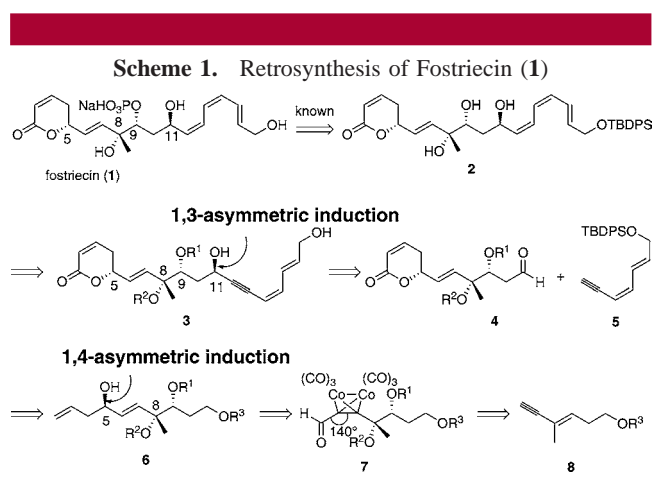
(3) Boger, D. L.; Ichikawa, S.; Zhong, W. *J. Am. Chem. Soc.* **2001**, *123*, 4161.

(4) (a) Chavez, D. E.; Jacobsen, E. N. *Angew. Chem., Int. Ed.* **2001**, *40*, 3667. (b) Reddy, Y. K.; Falck, J. R. *Org. Lett.* **2002**, *4*, 969. (c) Miyashita, K.; Ikejiri, M.; Kawasaki, H.; Maemura, S.; Imanishi, T. *Chem. Commun.* **2002**, 742. (d) Miyashita, K.; Ikejiri, M.; Kawasaki, H.; Maemura, S.; Imanishi, T. *J. Am. Chem. Soc.* **2003**, *125*, 8238. (e) Wang, Y. G.; Kobayashi, Y. *Org. Lett.* **2002**, *4*, 4615. (f) Esumi, T.; Okamoto, N.; Hatakeyama, S. *Chem. Commun.* **2002**, 3042. (g) Fujii, K.; Maki, K.; Kanai, M.; Shibasaki, M. *Org. Lett.* **2003**, *5*, 733. (h) Trost, B. M.; Frederiksen, M. U.; Papiillon, J. P. N.; Harrington, P. E. *J. Am. Chem. Soc.* **2005**, *127*, 3666. (i) Maki, K.; Motoki, R.; Fujii, K.; Kanai, M.; Kobayashi, T.; Tamura, S.; Shibasaki, M. *J. Am. Chem. Soc.* **2005**, *127*, 17111. (j) Yadav, J. S.; Prathap, I.; Tadi, B. P. *Tetrahedron Lett.* **2006**, *47*, 3773. Reviews, see: (k) Shibasaki, M.; Kanai, M. *Heterocycles* **2005**, *66*, 727. (l) Miyashita, K.; Ikejiri, M.; Tsunemi, T.; Matsumoto, A.; Imanishi, T. *J. Syn. Org. Chem. Jpn.* **2007**, *65*, 874.

(5) (a) Just, G.; Oconnor, B. *Tetrahedron Lett.* **1988**, *29*, 753. (b) Liu, S. Y.; Huang, D. F.; Huang, H. H.; Huang, L. *Chin. Chem. Lett.* **2000**, *11*, 957. (c) Cossy, J.; Pradaux, F.; BouzBouz, S. *Org. Lett.* **2001**, *3*, 2233. (d) Kiyotsuka, Y.; Igarashi, J.; Kobayashi, Y. *Tetrahedron Lett.* **2002**, *43*, 2725. (e) Marshall, J. A.; Bourbeau, M. P. *Org. Lett.* **2003**, *5*, 3197. (f) Ramachandran, P. V.; Liu, H. P.; Reddy, M. V. R.; Brown, H. C. *Org. Lett.* **2003**, *5*, 3755.

and three of the four chiral centers are derived from this chiral source.<sup>4j</sup>

Our synthetic strategy has been to control stereochemistry using pre-existing stereogenic centers as much as possible. Thus, the stereogenic center at C11 was to be constructed via 1,3-asymmetric induction using the C9 chiral center, whereas the stereochemistry at C5 was to be controlled by the chiral center at C8 via 1,4-asymmetric induction using a cobalt-alkyne complex, methodology recently developed in this group.<sup>6</sup> The only reaction requiring use of an external chiral auxiliary is the asymmetric dihydroxylation of a homoallylic alcohol. Another noteworthy feature is the construction of the labile triene unit, which we planned to synthesize by reduction of the more stable dieneyne at a late stage of the synthesis. Herein we report the realization of this scenario for the synthesis of protected dephosphofostriecin **2**, Imanishi's key intermediate (Scheme 1).<sup>4c,d</sup>



Our synthesis (Scheme 2) commenced with the preparation of cobalt-alkyne complex **7**. 1,3-Propanediol was mono-protected with PMBCl to give **10**, which was oxidized with  $\text{SO}_3 \cdot \text{pyridine}$ <sup>7</sup> to give the aldehyde. Wittig reaction, followed by reduction and oxidation, gave **13**. The Corey-Fuchs alkyne synthesis<sup>8</sup> afforded alkyne **14**, which on Sharpless dihydroxylation<sup>9</sup> gave diol **15** in good yield with excellent enantioselectivity (93% ee).<sup>10</sup> Protection of the diol **15** with TESCl gave **16**. Treatment of **16** with BuLi and  $\text{ClCO}_2\text{Me}$  gave ester **17**. Changing the protecting group from TES to BOM gave **19**. Reduction with DIBAL-H, followed by treatment with TBAF, gave diol **21**. Oxidation with  $\text{MnO}_2$  then gave aldehyde **22**. On treatment with  $\text{Co}_2(\text{CO})_8$  **22** provided cobalt-alkyne complex **23** in good yield.

(6) Hayashi, Y.; Yamaguchi, H.; Toyoshima, M.; Nasu, S.; Ochiai, K.; Shoji, M. *Organometallics* **2008**, *27*, 163.

(7) Parikh, J. R.; Doering, W. V. E. *J. Am. Chem. Soc.* **1967**, *89*, 5505.

(8) Corey, E. J.; Fuchs, P. L. *Tetrahedron Lett.* **1972**, *13*, 3769.

(9) Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. *Chem. Rev.* **1994**, *94*, 2483.

(10) The enantioselectivity of **15** was determined by chiral HPLC analysis (chiralcel OD-H column,  $l = 254 \text{ nm}$ ,  $i\text{PrOH}/\text{hexane } 1/20$ ,  $1.0 \text{ mL min}^{-1}$ ;  $t_R = 24.1 \text{ min}$  (major),  $20.7 \text{ min}$  (minor)).

Next is the critical 1,4-asymmetric induction. As allyl metal reagents such as allylMgBr gave no selectivity, the Hosomi-Sakurai allylation was investigated, with the results summarized in Table 1. Lewis acid mediated allylation<sup>11</sup> and Mukaiyama aldol reaction<sup>12</sup> of dicarbonylhexacarbonyl cobalt of  $\alpha,\beta$ -acetylenic aldehyde is known, and  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  is reported to be a suitable promoter in the reaction of allyl stannane.<sup>11</sup> Although low selectivity was observed with  $\text{HfCl}_4$ ,  $\text{ZnCl}_2$ ,  $\text{TiCl}_3(\text{O-}i\text{-Pr})$ ,  $\text{TiCl}(\text{O-}i\text{-Pr})_3$  and  $\text{BF}_3 \cdot \text{OEt}_2$ , moderate selectivity (73:27) was obtained when  $\text{MgBr}_2 \cdot \text{OEt}_2$  was employed. Good selectivity (80:20) was obtained in the presence of  $\text{TiCl}_2(\text{O-}i\text{-Pr})_2$ , which increased to 92:8 when the reaction was performed at lower temperature ( $-40 \text{ }^\circ\text{C}$ ). The bulkiness of the allyltin reagent is important, because  $\text{Ph}_3\text{SnCH}_2\text{CH}=\text{CH}_2$  gave a excellent result whereas the corresponding  $\text{Bu}_3\text{Sn}$  analogue gave poor selectivity (entry 9). The relative configuration of newly generated C5 of **24** was determined by the advanced Mosher's MTPA method<sup>13</sup> after conversion to (*R*)- and (*S*)-mono MTPA esters of **25**.

No selectivity was observed using the parent aldehyde **22** under the same reaction conditions. Thus, the cobalt-alkyne complex is the key for achieving high selectivity.

The angle of the alkyne triple bond is  $180^\circ$ , whereas that of the alkyne cobalt complex is about  $140^\circ$  (see **7** of Scheme 1). Complexation forces the stereogenic and prestereogenic centers closer, which makes the highly stereoselective 1,4-asymmetric induction possible. C5 stereochemistry has been constructed by the reagent-controlled allylation;  $\text{Ipc}_2\text{B-allyl}$ <sup>14</sup> was used by Falck,<sup>4b</sup> Hatakeyama,<sup>4f</sup> Trost,<sup>4h</sup> and Yadav,<sup>4j</sup> whereas Shibasaki<sup>4g,i,k</sup> used the catalytic method of Yamamoto allylation.<sup>15</sup>

As the crucial 1,4-asymmetric induction had been successfully achieved, the cobalt was removed by treatment with NMO, then reduction with Red-Al gave *trans*-alkene **26** in good yield. The unsaturated lactone moiety was constructed by acylation with acryloyl chloride, followed by exposure to Grubbs' second generation catalyst,<sup>16</sup> affording **28** quantitatively.

The next task is installation of the triene moiety via 1,3-asymmetric induction. The protecting group of the diol unit and nucleophile were found to be important for obtaining high selectivity. Cleavage of the BOM group afforded the diol **29**, which is Hatakeyama's intermediate.<sup>4f</sup> The diol **29**

(11) Balduzzi, S.; Brook, M. A.; McGlinchey, M. J. *Organometallics* **2005**, *24*, 2617.

(12) Review, see: Mukai, C.; Hanaoka, M. *Synlett* **1996**, 11.

(13) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **1991**, *113*, 4092.

(14) Brown, H. C.; Jadhav, P. K. *J. Am. Chem. Soc.* **1983**, *105*, 2092.

(15) Yanagisawa, A.; Kageyama, H.; Nakatsuka, Y.; Asakawa, K.; Matsumoto, Y.; Yamamoto, H. *Angew. Chem., Int. Ed.* **1999**, *38*, 3701.

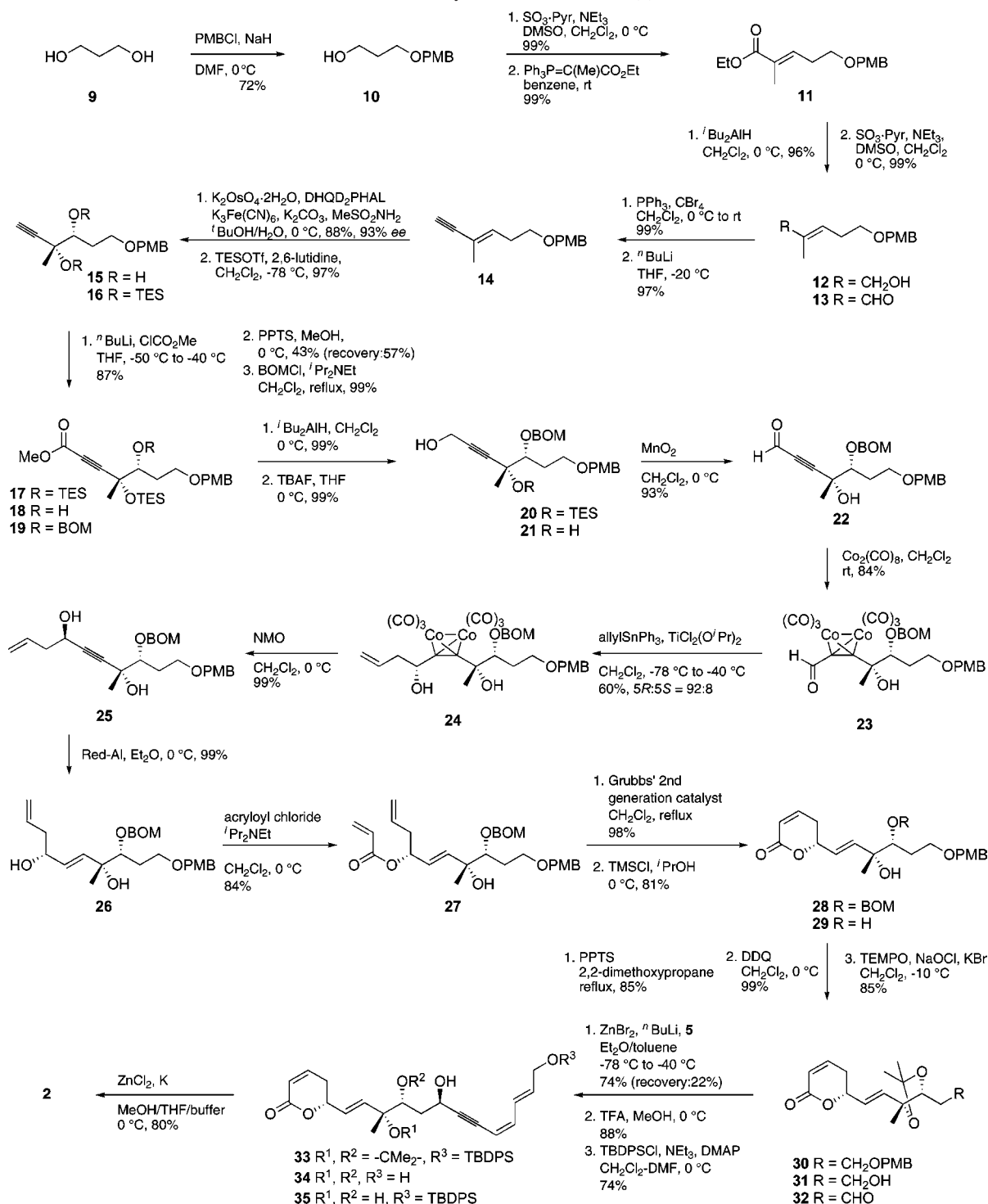
(16) (a) Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. *Org. Lett.* **1999**, *1*, 953. (b) Trnka, T. M.; Grubbs, R. H. *Acc. Chem. Res.* **2001**, *34*, 18.

(17) (a) Chan, K.-K.; Specian, A. C., Jr.; Saucy, G. *J. Org. Chem.* **1978**, *43*, 3435. (b) Jones, T. K.; Denmark, S. E. *Org. Synth. Col.* **7**, 524.

(18) Shimada, K.; Kaburagi, Y.; Fukuyama, T. *J. Am. Chem. Soc.* **2003**, *123*, 4048.

(19) For an excellent review of 1,3-asymmetric induction, see: Evans, D. A.; Dart, M. J.; Duffy, J. L.; Yang, M. G. *J. Am. Chem. Soc.* **1996**, *118*, 4322.

**Scheme 2. Synthesis of Fostriecin (1)**



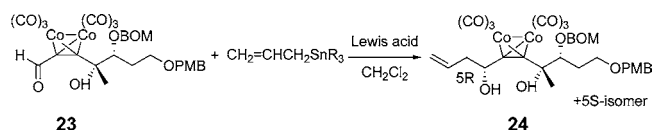
was protected as its dimethylacetal **30**. Removal of the PMB protecting group, followed by TEMPO oxidation, afforded key intermediate **32**.

The other coupling partner **5** was easily prepared stereoselectively from *cis*-1,2-dichloroethene via successive Sonogashira reactions (Scheme 3). That is, *cis*-1,2-dichloroethene (**36**) was reacted with 2-propyne-1-ol and tri-

methylsilylacetylene successively to generate ene diyne derivative **38** in good yield. Stereoselective reduction with Red-Al,<sup>17</sup> followed by protection of hydroxy group and deprotection of trimethylsilyl group, afforded **5** in good yield.

(20) (a) Rieke, R. D.; Uhm, S. J. *Synthesis* **1975**, 452. (b) Chou, W. N.; Clark, D. L.; White, J. B. *Tetrahedron Lett.* **1991**, 32, 299.

**Table 1.** Effect of the Lewis Acid on 1,4-Asymmetric Induction with **23** Using Allyltin



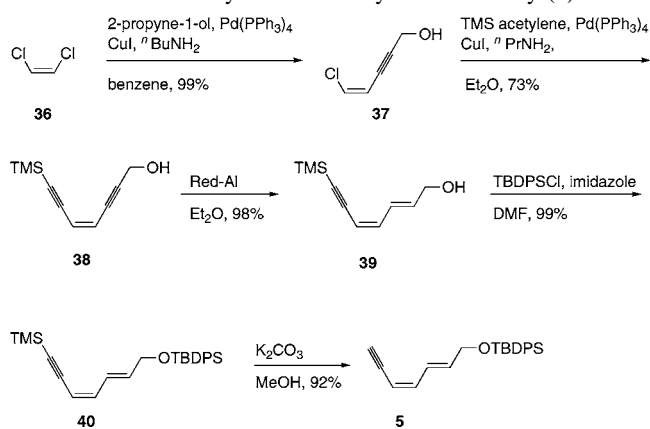
entry	R	lewis acid	condition	yield[%] <sup>a</sup>	R:S <sup>b</sup>
1	Ph	HfCl <sub>4</sub>	-78 °C, 30 m	15	50:50
2	Ph	ZnCl <sub>2</sub>	-78 °C, 4 h	45	55:45
3	Ph	BF <sub>3</sub> ·OEt <sub>2</sub>	-78 °C, 10 m	40	64:36
4	Ph	MgBr <sub>2</sub> ·OEt <sub>2</sub>	-78 ~ 0 °C, 8 h	86	73:27
5	Ph	TiCl <sub>3</sub> (O- <i>i</i> -Pr)	-78 °C, 30 m	20	60:40
6	Ph	TiCl(O- <i>i</i> -Pr) <sub>3</sub>	-78 ~ 0 °C, 8 h	78	60:40
7	Ph	TiCl <sub>2</sub> (O- <i>i</i> -Pr) <sub>2</sub>	-20 °C, 4 h	60	80:20
8	Ph	TiCl <sub>2</sub> (O- <i>i</i> -Pr) <sub>2</sub>	-40 °C, 46 h	60	92:8
9	Bu	TiCl <sub>2</sub> (O- <i>i</i> -Pr) <sub>2</sub>	-45 °C, 12 h	86	45:55

<sup>a</sup> Isolated yield. <sup>b</sup> Diastereomer ratio of C5 of **24** was determined by HPLC analysis.

After extensive experimentation, the diastereoselective coupling of **32** and **5** was successfully accomplished using Fukuyama's protocol.<sup>18,19</sup> The higher order alkynyl Zn reagent reacted smoothly to yield **33** in good yield with excellent diastereoselectivity. Treatment of **33** with TFA removed both the acetal and TBDPS groups to yield tetraol **34**. Protection of the primary alcohol with TBDPSCl gave **35**. The labile triene unit was successfully constructed using Rieke Zn<sup>20</sup> in the presence of buffer solution, affording the protected dephosphofostriecin **2** in good yield. The conversion of **2** to fostriecin has been demonstrated by Imanishi and co-workers.<sup>4c, d</sup>

In summary, we have accomplished the synthesis of a protected dephosphofostriecin, and thereby a formal synthesis of fostriecin. Two of the four chiral centers are

**Scheme 3.** Synthesis of Alkynediene Moiety (**5**)



controlled by an external chiral reagent and the other two are synthesized stereoselectively, one by a novel 1,4-asymmetric induction using cobalt-alkyne complex developed by our group<sup>6</sup> and the second by 1,3-asymmetric induction.

**Acknowledgment.** This work was partially supported by a Grand-in-Aid for Scientific Research on Priority Areas 16073219 from MEXT.

**Supporting Information Available:** Detailed experimental procedures, full characterization, and copies of <sup>1</sup>H and <sup>13</sup>C NMR and IR spectra of all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL800195G